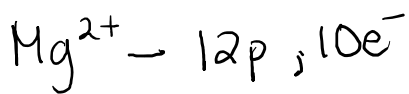
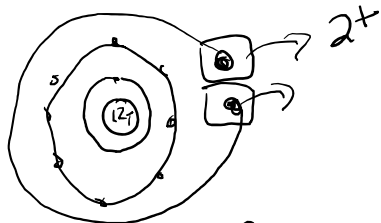
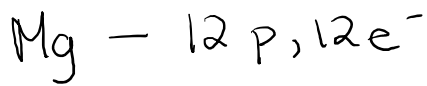


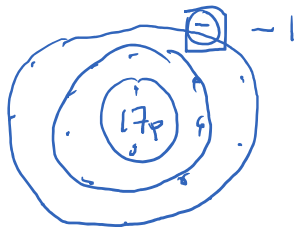
Review Class

Friday, May 5, 2017 12:58 PM

Ion Size



→ cations are smaller than their neutral element



→ anions are larger than their neutral element



ΔEN

ΔEN

∅

L.7

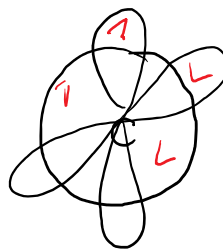
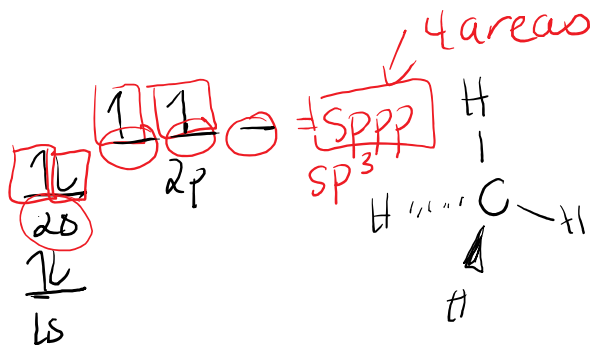
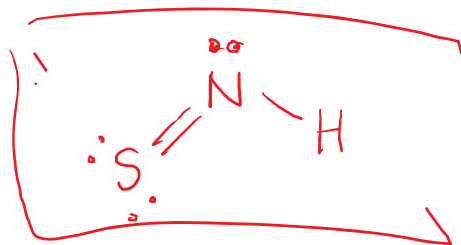
pure covalent

polar covalent

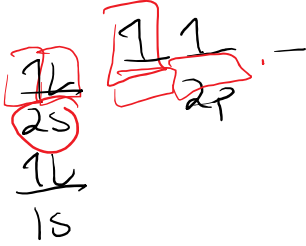
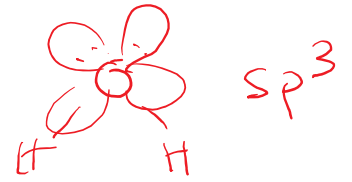
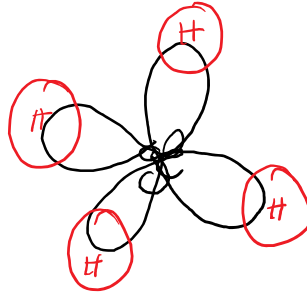
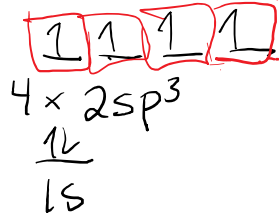
ionic

(non-polar)

∴ diatomic



ls



double bond
 - 1 sigma
 - 1 pi

